



# POROUS POWDER MATERIALS

SYMPOSIUM & EXHIBITION



3-6

September

#### **MAIN THEMES:**

Theme A: Development and Characterization

**Theme B:** Catalytic Aspects

Theme C: Environmental and Hygenic Aspects
Theme D: Biological and Medical Aspects

Theme E: Transport and Surface Chemistry

Theme F: Modeling and SimulationTheme G: Industrial Applications

**Proceedings:** Full papers submitted will be printed in the Symposium Proceedings Book.

#### **VENUE**

The symposium will be held in Sheraton Hotel Resort & Spa in Izmir-Cesme-Turkey. Register now and enjoy the low registration and hotel rates.

## organizing **COMMITEE**

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#### **INVITED SPEAKERS**

Ned ALLEN / Lockheed Martin Corp., USA

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Paolo COLOMBO / Univ. of Padova, ITALY

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Muharrem KAPLAN / Adacal Industrial Minerals, TURKEY

Carl LEVOGUER / Malvern Instruments Ltd., UK

Jan D. MILLER / Univ. of Utah, USA

José RIVAS / Int'l Iberian Nanotech. Lab., PORTUGAL

Lev SARKISOV / Univ. of Edinburgh, UK

Matthias THOMMES / Quantachrome Corp., USA S. Ali TUNCEL / Hacettepe Univ., TURKEY

Kemal YILDIRIM / Dedeman Minimng Group, TURKEY

Symposium's offical Organizing agency is ALBEDO Tourism



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	POSTER PRESENTATIONS (Updated 25/08/2013)						
ID	Title	Name	Institution	Country			
5	POROUS SILICON: Determining the Critical Anodisation Current	Şafak Doğan	gazi university	Turkey			
10	Theoretical study of thiol-silica-cisplatin interactions	Andres Diaz Compañy	UNS, UTN	Argentina			
11	Sulfurization of activated carbon with sulfur dioxide in a fluidized-bed furnace	Tahereh Kaghazchi	Amirkabir University of Iran	Iran			
12	Microstructure and characterization of B4C reinforced Al2024 alloy matrix composites produced by mechanical alloying technique	Aykut Canakci	Karadeniz Technical University	Turkey			
13	Artificial neural network approach to predict of effect of process control agent on the microhardness of Al-Al2O3 composite powders produced by mechanical alloying	Aykut Canakci	Karadeniz Technical University	Turkey			
15	Fabrication of Fe–Al intermetallic coatings using mechanical milling technique	Aykut Canakcı	Karadeniz Tecnical Universty	Turkey			
16	Abrasive wear behavior of powder metallurgy CuSn10-graphite metal matrix composites produced by powder metallurgy	Aykut Canakci	Karadeniz Technical University	Turkey			
17	Application of Oxidized and Reduced Activated Carbon for para- nitrophenol removal	Tahereh Kaghazchi	Amirkabir university of technology	Iran			
19	EFFECT OF THE BINARY AND TERNARY EXCHANGES ON CRYSTALLYNITY AND TEXTURAL PROPERTIES OF X ZEOLITES	HABIB HAMMOUDI	UNIVERSITY OF MOSTAGANEM	Algeria			
22	Isomerisation of paraffins nC10-nC17 OVER Pt/H-Halloysites	Ahmed BELHAKEM	Dept Chemistry, F.S.E&I/U. Mostaganem	Algeria			
23	Adsorption studies of lead ions from wastewater using waste foundry sand	Semra Çoruh	Ondokuz Mayıs Üniversity, Departement of Environmental Eng.	Turkey			
24	Relation Between Rheological Properties and Pore Structure of the Fly ash/Cement Based AAC Mortar and Product Properties	M.Serhat Başpınar	Afyon Kocatepe University	Turkey			
27	Study of desorption in a saturated clay with nitrogen of medium fraction of petroleum	Marco Figueiredo	Universidade do Estado do Rio de Janeiro	Brazil			
28	Influence of Surface Chemistry on β-galactosidase from Kluyveromyces Lactis immobilisation onto Cellulose Acetate Microfiltration Membrane	Hacı Ali GÜLEÇ	Trakya University	Turkey			
29	Hydrogen Storage on M1+-ZSM-5 Clusters (M = K, Li and Na)	Mehmet Ferdi FELLAH	Bursa Technical University	Turkey			
30	Effect of pore wall microstructure on the mechanical properties of low alloy steel foams	Nuray Beköz	Istanbul University, Engineering Faculty, Metallurgical and Materials Engineering Department	Turkey			
32	Mechanical properties of sinter-hardened Cu-Ni-Mo based steels	Nuray Beköz	Istanbul University, Engineering Faculty, Metallurgical and Materials Engineering Department	Turkey			
33	The effect of the porous Bentonite-AlCrNi material in the turbine efficiency preservation	Mokhtar BOUNAZEF	Djillali Liabes University of Sidi Bel Abbes	Algeria			
36	Analysis of Swelling Behaviour of Clay Minerals by Discrete Numerical Simulation	hamed bayesteh	Univercity of Tehran	Iran			
40	Loaded polymer foam material for microwave absorption	Ratiba BENZERGA	IETR	France			
41	Microwave absorbing properties of reinforced foam-glasses designed for building industry	Ratiba BENZERGA	IETR	France			
43	Simulation of Micro-permeability during solidification of Al-3%Si mushy alloy as a dynamic porous media	sadegh firoozi	amirkabir University	Iran			
44	Kinetic sudy of major components and antibacterial activity of the	Nafila Zouaghi	Ecole Normale Supérieure Kouba	Algeria			
45	Electromagnetic wave absorption properties of Magnetic Nanowires Growth in Nanoporous Anodic Alumina Template	Majid Ebrahimzadeh	Behin Sanaat Yam Company, Saac	Iran			
46	THE EFFECT OF FLY ASH AND MICRO – SILICA ON CONCRETE QUALITY IN PRODUCTION OF CONCRETE	SULEYMAN ULUOZ	ILGAZ INSAAT	Turkey			
47	The role of the zeolite structure on the catalytic performance in gas- phase phenol oxidation by N2O	Larisa Pirutko	Boreskov Institute of Catalysis	Russian Federation			

485	Synthesis and characterization of SiC / Al MMCs by powder metallurgy	Onur Fevzi	Kırıkkale University	Turkey
	Synergistic effects of HPC and SDS on fast redispersion and dissolution of	KEVENLİK	New Jersey Institute of	•
486	fluidized bed and spray dried drug nano-composite powders	Rajesh Dave	Technology	United States
487	Atmospheric pressure synthesis of mesoporous silica (MCM-41 and HMS) for catalytic application in bidiesel production	Tahere Mohamadpanah	Materials and energy research center	Iran
489	Biodegradation of Phenol by pseudomonas spp	zoubida bendjama	laboratoire sciences du génie des procédés, faculté de génie mécanique et génie des procédés,USTHB,	Algeria
491	Effect of flow field and geometry of regular scaffolds in cell culture parameters	Javad Etedadi	Sahand University of Technology, Mechanical Engineering Department, Biomechanics Group	Iran
492	Microporous polymer networks (MPNs) made in metal-free regimes – systematic optimization of a synthetic protocol towards N-arylcarbazole-based MPNs	Eduard Preis	Bergische Universität Wuppertal	Germany
495	Metal-Organic Frameworks as Sensor materials for selective Detection of Guest Molecules	Nicole Klein	Fraunhofer IWS Dresden	Germany
496	Characterization and Modification of Kaolinite with Pyridine and 2,6- Diamino Pyridine	BERNA KOÇER	BALIKESIR UNIVERSITY	Turkey
497	Monodisperse Porous Polymer Particles Containing N-Methyl-D- Glucamine Groups for Boron Removal from Geothermal Water by Hybrid Method	NALAN KABAY	Ege University	Turkey
499	Molecular Dynamics Simulation of Graphene Like Boron Carbide Nanoribbons Under Strain	Yusuf Şimşek	Vocational School of Health Service, Gazi University, Ankara	Turkey
500	Development of a New Nondestructive Method for Monitoring the Homogeneity of Pharmaceutical Powders	NADJET GUEMRAS	USTHB/CINaM-CNRS AIX MARSEILLE UNIVERSITY	Algeria
502	The Effect of Cooling Rate on the Characteristics of Unidirectionally Freeze-Casted Gelatin Scaffold	Ali Zamanian	Materials and Energy Research Center	Iran
505	Annealing of Co-Cr dental alloy: Effects on nanostructure and Rockwell hardness	Simel Ayyildiz	Department of Prosthodontics, Dental Health Sciences Center, Gülhane Military Medical Academy	Turkey
506	EFFECTS OF B4C ON ALUMINA-BASED CERAMIC FOAM FILTERS	Kerim Emre ÖKSÜZ	Cumhuriyet University	Turkey
508	The Effect of Composition and Cooling rate on Microstructure of Chitosan/Silk-fibroin Microtubular Ice-tamplated Porous Scaffold	Ali Zamanian	Materials and Energy Research Center	Iran
509	The Photocatalytic Activity of Ceria Doped Titania Catalysts in Methylene Blue Degradation and Glycerol Reforming	Mert Tunçer	Izmir Institute of Technology, Department of Chemical Engineering	Turkey
510	Improvement of HA bioceramics with addition of AW glass-ceramic	Fatih ÇALIŞKAN	Sakarya University	Turkey
511	Formulation and characterization of the microparticles containing the antihypertenseurs for pharmaceutical use	NADJET GUEMRAS	USTHB/CINAM-CNRS/AIX MARSEILLE UNIVERSITY/CRAPC	Algeria
513	The effect of waste marble powder on the fresh state and mechanical properties of self-compacting mortars	Çağlar Yalçınkaya	Dokuz Eylül University, Department of Civil Engineering	Turkey
515	Characterization and Incresing Porosity of Limestones;Kufeki	Sonmez Arslan	Batman University	Turkey
516	Use of microorganisms immobilized on graphite to remove lead from aqueous solution	Hassiba MOKADDEM	USTHB	Algeria
517	Recovery of isoamyl acetate from aqueous solution using macroporous adsorption resins	Murat Yılmaztekin	Inonu University	Turkey
518	Removal of mercury from aqueous solutions using composite nanomaterials of graphite oxide with chitosan and magnetic chitosan	Eleni Deliyanni	Laboratory of General & Inorganic Chemical Technology, Division of Chemical Technology, School of Chemistry, Aristotle University of Thessaloniki	Greece



# Removal of mercury from aqueous solutions using composite nanomaterials of graphite with chitosan and magnetic chitosan

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ABSTRACT—In the current work, two novel forms of graphite oxide composites were prepared having as substrate chitosan. The first composite (GO-Ch) is a complex between graphite oxide (GO) and cross-linked chitosan (Ch), while the other composite (GO-Chm) is consisted of GO and magnetic chitosan (Chm). The magnetism occurred in the second case can strongly influence the properties of two polymers. A full characterization was achieved in order to study the properties of the composites synthesized. X-ray diffraction patterns revealed the crystallinity and possible degree of order. The elemental analysis of materials evaluated using energy dispersive X-ray analysis (EDX), while their morphology was observed with SEM. Also, TG analysis was realized revealing a significant difference in thermal behavior between the composites and the separate parts (GO and Ch/Chm). FTIR spectroscopy was used in order to explain, via the shift of peaks, the new interactions occurred in the composites. Then, the composite materials prepared used in adsorption experiments to bind/remove Hg(II). The maximum adsorption capacity can reach 190 mg/g (GO-Ch) and 210 mg/g (GO-Chm) for the removal of mercury. Kinetics as well as desorption experiments were done. Five cycles of adsorption-desorption were carried out.

### Materials preparation

GO was prepared according to Hummers method

CS was cross-linked with GLU

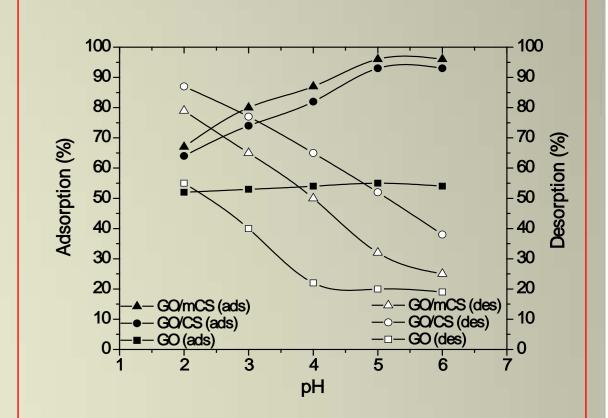
GO-C/CSm composite: CS/mCS solution (2% w/v) was prepared by dissolving of chitosan into acetic acid solution (2% v/v) under ultrasonic stirring for 2 h at room temperature .GLA were added to cross-linked chitosan. GO were added in the solution prepared and the mixed system was stirred for 90 min in a water bath at 50 °C and further 60 min at 80 °C. Black products were washed with ethanol and distilled water and dried in a vacuum oven at 50 °C.

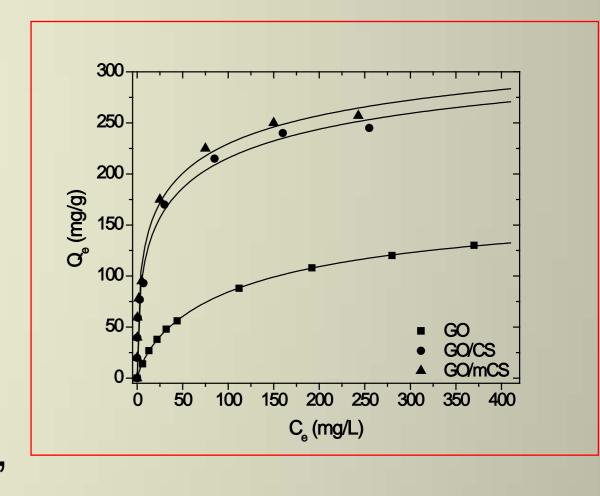
GO/CS and GO/CS and GO/CS and GO/CS and GO/CS was drastically influenced from pH conditions, while that of GO was unchanged.

Qmax of the initial material of GO (187 mg/g) was increased to 381 mg/g (GO/CS) and 397 mg/g (GO/mCS) mg/g. Increase of Qmax for all adsorbents with the increase

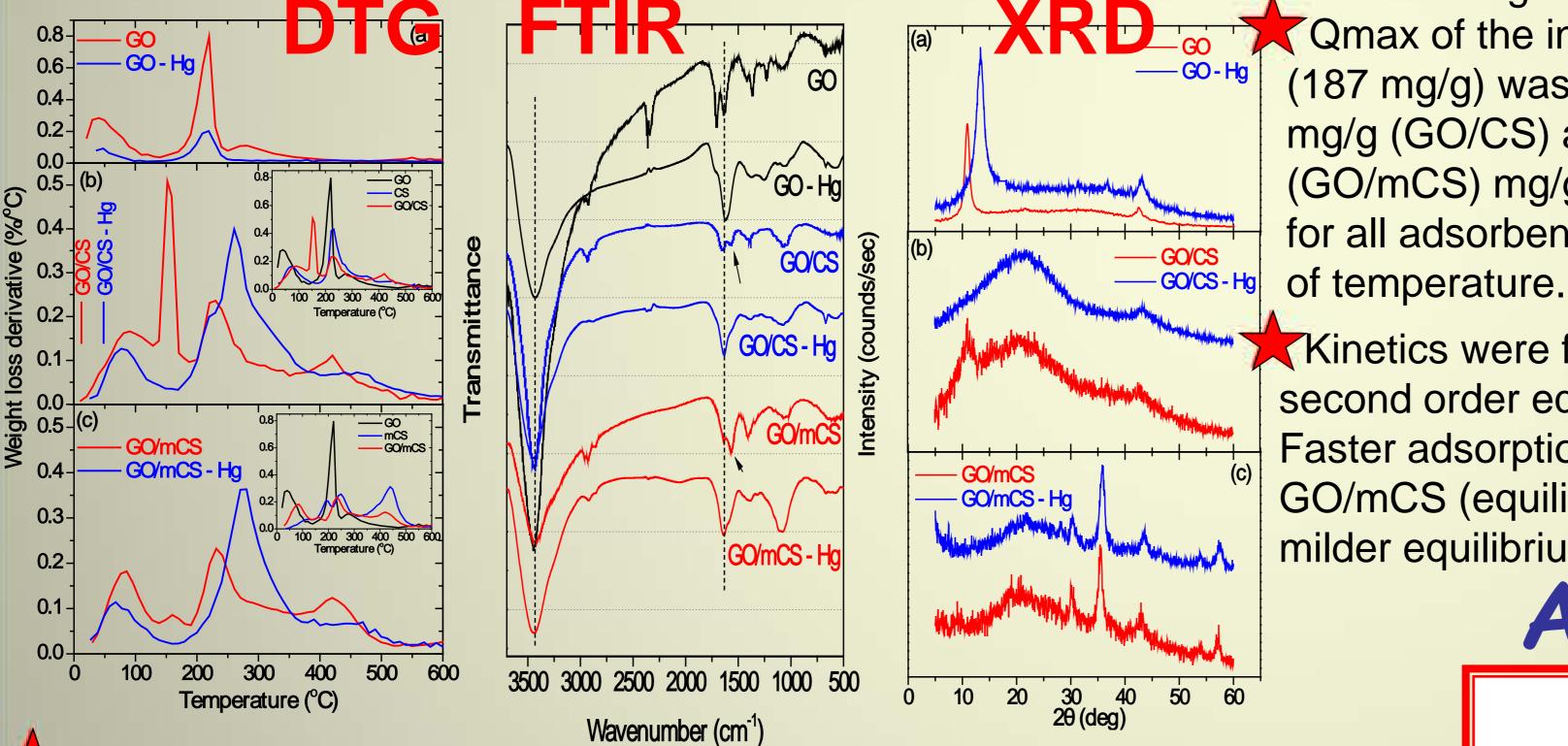
Kinetics were fitted to the pseudo–second order equation (R2~0.999). Faster adsorption for GO/CS anf GO/mCS (equilibrium after ~150 min), milder equilibrium

## Adsorption results



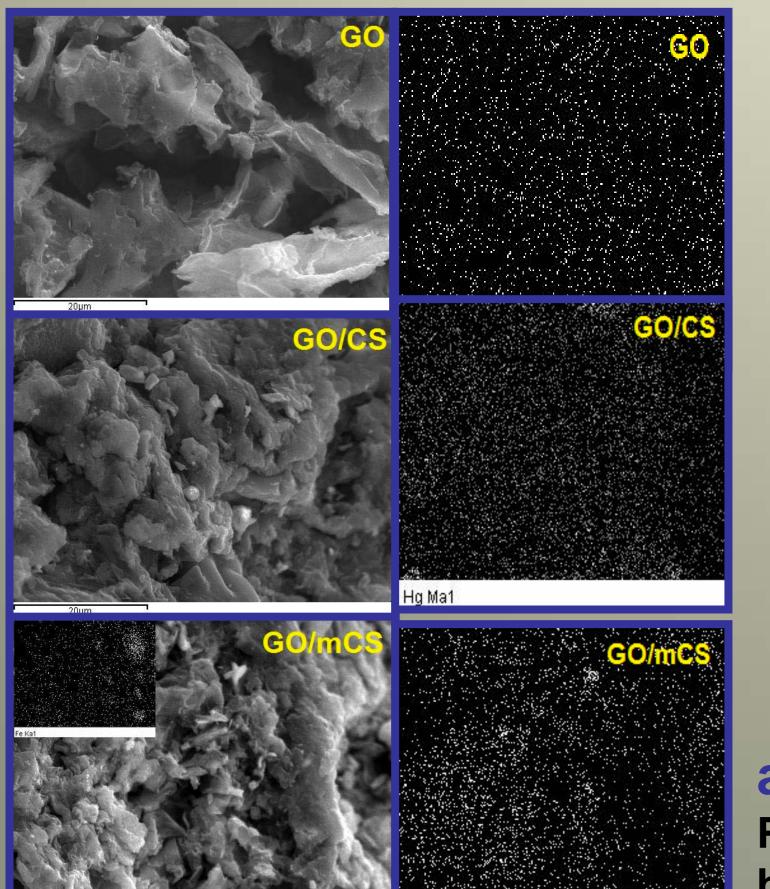


Materials characterization



After Hg(II) adsorption onto GO, a decrease in the interlayer distance between carbon layers was observed, in line with the presence of

Hg0 onto surface.



Hg(II) onto GO/CS. The exfoliation/destruction of the layered structure of GO during synthesis of GO/mCS indicating that carboxyl and epoxy groups between GO layers reacted with a significant part of the amino groups of mCS containing magnetic nanoparticles with a size bigger than that of its interlayer distance. After Hg(II) onto GO/mCS no other structural change was occurred.

Adsorption mechanism

Adsorption

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Proposed Hg(II) adsorption mechanism (via characterization and adsorption evaluation): chelation between amino groups of CS and Hg(II). Possible interactions between oxygen atoms from carboxyl or hydroxyl groups of GO-based material and Hg(II) (especially at pH conditions where chelation does not favor). adsorption mechanism (via characterization and adsorption evaluation):