

THESSALONIKI, AUGUST 26-29, 2018

Under the aegis of the Ministry of Environment and Energy

**CONFERENCE PROGRAM** 

### POSTER PRESENTATIONS

### MONDAY, AUGUST 27, 2018

Morning session (11:00 - 14:00)
---------------------------------

Morning session (11:00 – 14:00)				
A01	DRY ANAEROBIC DIGESTION OF SEWAGE SLUDGE IN SMALL AND MEDIUM WASTEWATER TREATMENT PLANTS  L. Pastor, J.E. Sánchez- Ramírez, S. Doñate, E. Aymerich, M. Esteban- Gutiérrez, J. Garcia- Aguirre, R. Romaguera, C. Lardin and E. Mino	A12	PHOTOCATALYTIC REMOVAL OF PESTICIDES IN WATER BY TIO <sub>2</sub> /GRAPHENE OXIDE COMPOSITES  C. Berberidou, G. Z. Kyzas, and I. Poulios	
A02	SMALL AND DECENTRALIZED WASTEWATER TREATMENT PLANTS IN SMALL MUNICIPALITIES OF EAST CRETE  M. P. Kavvalakis and E. G. Dialynas	A13	"AQUASENSE": ADAPTIVE WATER CONSUMPTION MANAGEMENT SERVICE D. Chondrogianni, S.Karatzas and P.G.Saranti	
A03	STUDY ON THE TREATMENT OF ALCOHOL DISTILLERY WASTEWATER WITH ENERGY MICROALGAE  Zongxin Ran, Jiang Yu, Ye Du, Wei Wei, Weidong Peng, Yulong Hea and Yunxiao Peng	A14	REMEDIATION STUDY OF HERBICIDE terbuthylazine IN CONSTRUCTED WETLAND MICROCOSM PLANTED BY Typha latifolia L. N. Papadopoulos and G. Zalidis	
A04	INTELLIGENT SUSTAINABLE URBAN DRAINAGE SYSTEMS (I-SUDS): A FRAMEWORK FOR FLOOD MITIGATION AND RAINWATER REUSE S. Karatzas, D. Chondrogiani and P. Saranti	A15	EFFECTIVENESS OF PLANT GROWTH-PROMOTING RHIZOBACTERIA INOCULATION OF CUPRESSUS SEMPERVIRENS IRRIGATED BY TREATED WASTE WATER  M. Girgis and M. El- Sayed	
A05	OPTIMIZATION OF SILICA EXTRACTION FROM DIATOMACEOUS EARTH (DE) USING RESPONSE SURFACE METHODOLOGY (RSM) O Izevbekhai, W Gitari and N Tavengwa	A16	QUALITATIVE ASSESSMENT OF WATER PARAMETERS IN THE AREA OF THESSALONIKI  I. Giannakis, C. Emmanouil and A. Kungolos	
A06	NOBLE METAL MODIFIED TIO2 NANOPARTICLES FOR EFFICIENT PHOTOCATALYTIC PURIFICATION OF DRINKING WATER C. Berberidou and I. Poulios	A17	ANALYSIS OF SOME METALS IN THE SOLID-LIQUID PHASES FROM SOUTHEASTERN MEDITERRANEAN BASIN, EGYPT Maha Ahmed Mohamed Abdallah	
A07	ELIMINATION OF EMERGING POLLUTANTS WITH MEMBRANE PROCESSES: A REVIEW  J. Jaime Sadhwani Alonso, B. Del Río- Gamero and C. Méndez- Montes	A18	IMPROVING PERFORMANCES OF NANO-CARBON MEMBRANES WITH ELECTROCHEMICAL ASSISTANCE FOR WATER TREATMENT Xie Quan, Haiguang Zhang, Xinfei Fan	
A08	ENVIRONMENTAL SOLUTIONS AND DEODORIZATION APPLICATIONS WITH DIATOMACEOUS EARTH A.S. Dounavis, M. Nakou, A. Peleka, A. Maragkoulidis and X. Papanikos	A19	DIRECT RENEWABLE ELECTRICITY GENERATION FROM MUNICIPAL WASTEWATER SLUDGE SIMULTANEOUS WITH ITS DIGESTION BY A NOVEL TWO-CHAMBERED AEROBIC DIGESTER FOR SMALL COMMUNITIES H. Rasouli Sadabad, G. Badalians Gholikandi, M. Amouamouha	
A09	A COMPARATIVE STUDY OF THE CHARACTERISTICS OF RAW LEACHATES FROM TWO LANDFILL SITES (AN OPEN AND A CLOSED) IN THE THESSALONIKI PREFECTURE  A. I. Koumalas, A.S. Dounavis, E.M. P. Barampouti and S. Th. Mai	A20	SYNTHESIS AND CHARACTERIZATION OF MEMBRANE BASED ON CHITOSAN AND THEIR APPLICATION IN THE ENVIRONMENT: DISCOLORATION OF POLLUTED INDUSTRIAL WATERS H. Laribi- Habchi , L. Adour , W. Arbia, A. Hamdoune and W. Ghomaid	
A10	INDUSTRIAL SYMBIOSIS: FUR INDUSTRY WASTE UTILIZATION AS SECONDARY FUEL FOR SEWAGE SLUDGE DRYING (KOZANI WASTEWATER TREATMENT PLANT)  A. T. Zamanis, P. Barouchas and A. Zoumboulis	A21	ECO-SUSTAINABLE TEXTILE TREATMENTS TO MITIGATE THE RELEASE OF MICROPLASTICS IN THE WASTEWATER OF LAUNDRY PROCESSES F. De Falco, V. Guarino, G. Gentile, R. Avolio, M. E. Errico, V. Ambrogi, E. Di Pace, M. Avella and M. Cocca	
A11	CLOUD POINT EXTRACTION FOR ORGANICS ACIDS RECOVERY FROM TABLE OLIVE WASTEWATERS J. Raiti, A. El-Abbassi, H. Kiai, A. Hafidi	A22	SEWAGE CHALLENGE IN THE AREA OF ZGHARTA – NORTH OF LEBANON S. AbiDib Antoun	

### PHOTOCATALYTIC REMOVAL OF PESTICIDES IN WATER BY TIO<sub>2</sub>/GRAPHENEOXIDE COMPOSITES

C. Berberidou, G. Z. Kyzas, and I. Poulios

Department of Chemistry, Aristotle University of Thessaloniki, 54124, Thessaloniki, Greece Hephaestus Advanced Laboratory, Eastern Macedonia and Thrace Institute of Technology, Kavala, Greece

E-mail: cberber@chem.auth.gr

#### **ABSTRACT**

There is a huge body of evidence supporting the hypothesis of pesticides being one of the main environmental stressors and their direct correlation with diverse impacts on human health. Several studies have established the epidemiological association between pesticide exposure and elevated rate of chronic diseases such as different types of cancers, diabetes, neurodegenerative disorders like Parkinson and, Alzheimer, birth defects, and reproductive diseases. United Nations estimates that approximately 99% of pesticides used in agriculture contaminates soil, air and especially water, with unpredictable consequences. Indeed, alternative, environmentally friendly, yet, efficient methods of water purification, have become an urgent need. In this context, Advanced Oxidation Processes (AOPs) have been particularly effective in the decomposition of a wide range of pollutants present in various environmental media. Among them, heterogeneous photocatalytic oxidation in the presence of semiconducting oxides, mainly TiO<sub>2</sub>, has served as one of the most promising alternatives for drinking water processing. Due to its associated advantages, various modifications of the TiO<sub>2</sub> photocatalytic process have been suggested, including coupling of TiO<sub>2</sub> with graphene-based materials.

In the present study, preparation of graphene oxide and its' conjugation with  $TiO_2$  has been conducted, aiming to the enhancement of the photocatalytic efficiency of  $TiO_2$  and/or the exploitation of visible light spectrum. Various parameters have been investigated, including the amount of graphene oxide and the type of  $TiO_2$  employed in the synthesis of the  $TiO_2/GO$  composites. Clopyralid (3,6-dichloro-2-pyridine-carboxylic acid), a systemic herbicide from the chemical class of pyridine compounds, often detected in drinking water, served as a model pollutant for the evaluation of the photocatalytic efficiency of the prepared mixed oxides.

**Keywords**: clopyralid, graphene oxide, photocatalytic, TiO<sub>2</sub>.

### Acknowledgements

This research is implemented through the State Scholarships Foundation (IKY) program and co-financed by the European Union (European Social Fund-ESF) and Greek national funds through the action "Reinforcement of Postdoctoral Researchers", in the framework of the Operational Program "Human Resources Development Program, Education and Lifelong Learning" of the National Strategic Reference Framework (NSRF) 2014-2020.

94 | E- Proceedings of the 5th International Conference

on Small and Decentralized Water and Wastewater Treatment Plants





# PHOTOCATALYTIC REMOVAL OF PESTICIDES IN WATER BY

## TiO<sub>2</sub>/GRAPHENE OXIDE COMPOSITES



Chrysanthi Berberidou<sup>1</sup>, George Z. Kyzas<sup>2</sup> and Ioannis Poulios<sup>1</sup>

<sup>1</sup> Laboratory of Physical Chemistry, Department of Chemistry, Aristotle University of Thessaloniki, Thessaloniki, Greece
<sup>2</sup> Hephaestus Advanced Laboratory, Eastern Macedonia and Thrace Institute of Technology, Kavala, Greece

\* Correspondence: cberber@chem.auth.gr

## **Introduction and Objectives**

There is a huge body of evidence demonstrating that pesticides are, indeed, on top of the list of the main environmental stressors with diverse impacts on human health. According to United Nations, approximately 99% of pesticides used in agriculture contaminates soil, air and especially water, with unpredictable consequences.

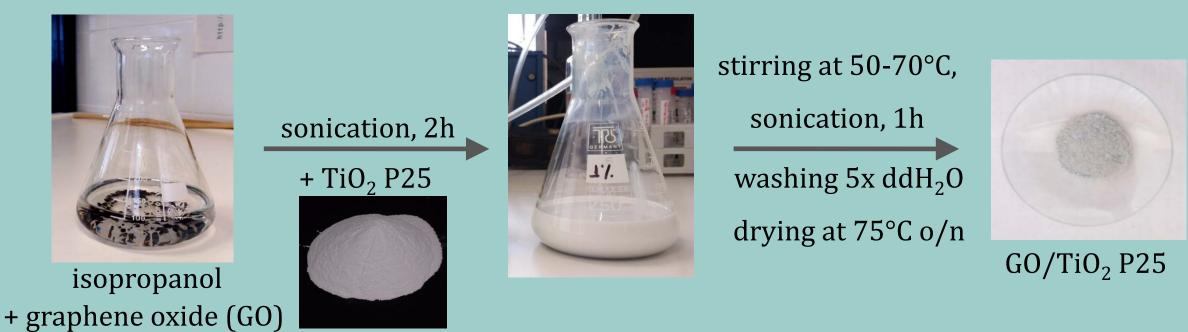
In this context, Advanced Oxidation Processes (AOPs) and  $TiO_2$  mediated photocatalytic oxidation, in particular, have been remarkably effective in the decomposition of a wide range of pollutants present in various environmental media. Due to its associated advantages, various modifications of the  $TiO_2$  photocatalytic process have been suggested, including coupling of  $TiO_2$  with graphene-based materials.

In the present study, preparation of graphene oxide (GO) and its' conjugation with  ${\rm TiO_2}$  has been conducted, aiming to the enhancement of the photocatalytic efficiency of  ${\rm TiO_2}$ . Clopyralid, a systemic herbicide from the chemical class of pyridine compounds, often detected in drinking water, served as a model pollutant for the evaluation of the photocatalytic efficiency of the prepared mixed oxides.

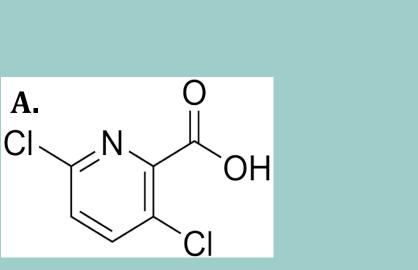
### **Materials and Methods**

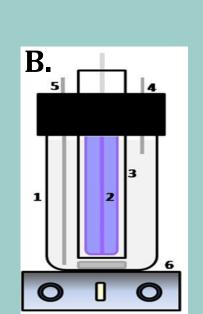
GO was synthesized with the modified Hummers method (Hummers and Offeman, 1958). Commercial graphite powder was stirred in the presence of concentrated  $\rm H_2SO_4$  at 0°C. KMnO<sub>4</sub> was slowly added to the suspension, at temperature below 20°C. The reaction mixture was cooled to 2°C and was then stirred at room temperature for 30 min. Distilled water was slowly added to the reaction vessel, keeping the solution temperature below 98°C. The suspension was stirred for additional 15 min, was further diluted with distilled water and 30% w/w  $\rm H_2O_2$  was added. The mixture was left overnight. Precipitated GO was separated by decantation. The remaining suspension was transferred to dialysis tubes. Dialysis was carried out until no precipitate of  $\rm BaSO_4$  was detected by addition of aqueous solution of  $\rm BaCl_2$ . Then, GO was separated by centrifugation. The gel-like material was freeze dried and a fine dark brown powder of GO was obtained.

 $\rm GO/TiO_2$  mixed oxides were prepared as shown in Fig. 1. The desired amount of GO was added to isopropanol and was sonicated for 2h.  $\rm TiO_2$  P25 (Evonik) was added to the exfoliated GO and was stirred for 2h at 50-70 °C. The suspension was then sonicated for 1h and was then , dried at 75 °C o/n. The mixed oxide was then left to cool at RT, grinned and washed 5 times with doubly distilled water. The final product, a grey powder, was obtained after drying at 75 °C o/n.



**Figure 1:** Route of preparation of graphene oxide/TiO<sub>2</sub> P25 composites.

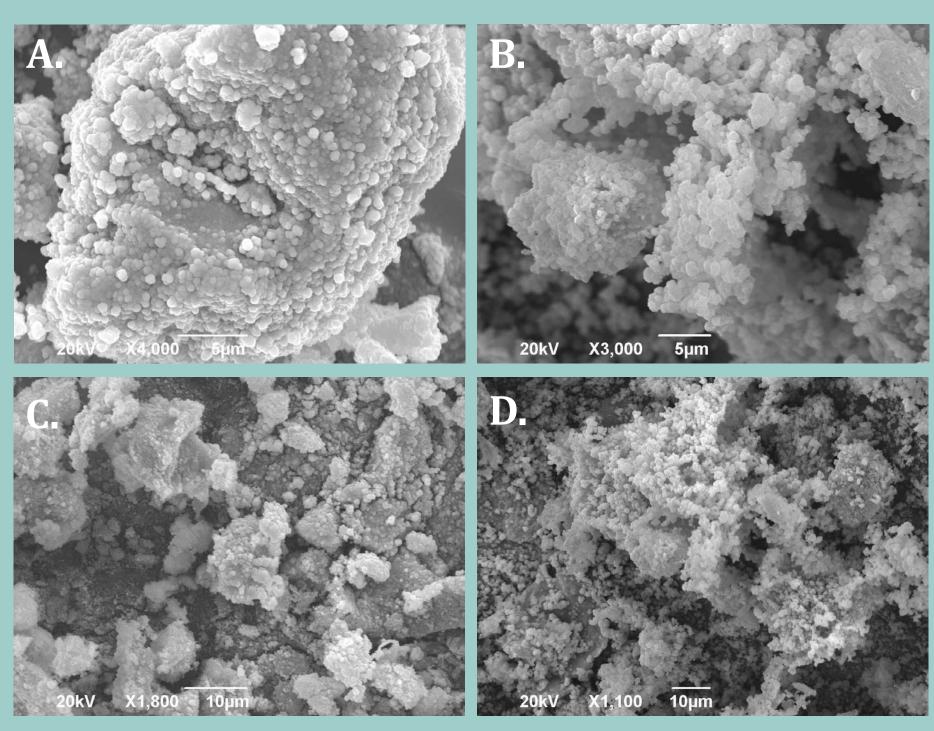


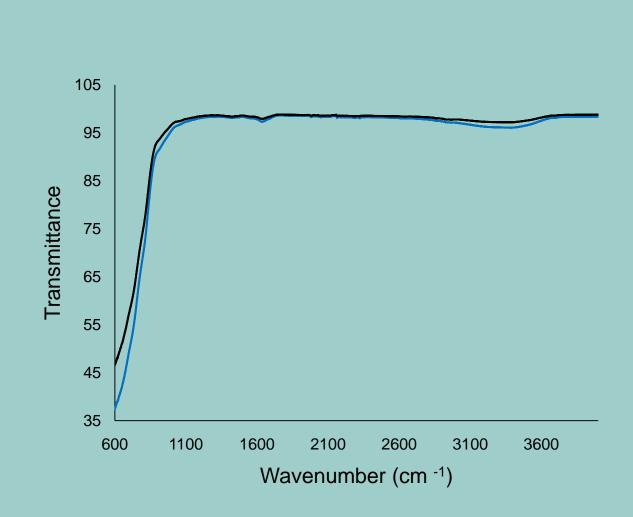


**Figure 2 A**: Clopyralid (3,6-dichloro-2-pyridine-carboxylic acid, CAS No. 1702-17-6, Mr: 192 g mol<sup>-1</sup>). **B**: Sketch of the photocatalytic reactor employed in the photocatalytic oxidation of clopyralid: 1. reaction vessel, 2. irradiation source, 3. borosilicate glass, 4. air inlet, 5. air outlet, 6. stirrer.

Photocatalytic oxidation of clopyralid was performed in a closed Pyrex cell of 300 mL capacity, fitted with a central Osram Dulux 9W/78 lamp, under magnetic stirring (Fig. 2B). The cell had inlet and outlet ports for bubbling  $CO_2$  free air. Photocatalysis was conducted at a working volume of 250 mL, at  $3.8\pm0.1$  initial pH and constant temperature ( $25\pm0.1^{\circ}C$ ).

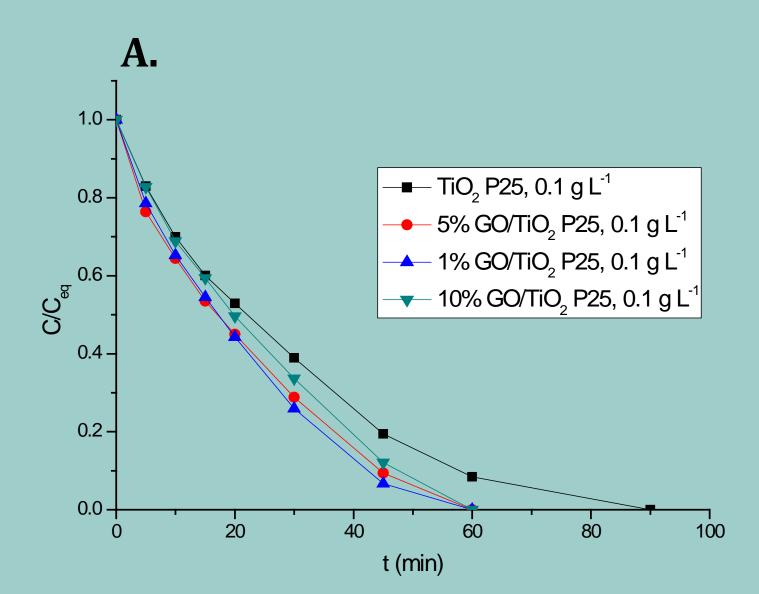
## Results

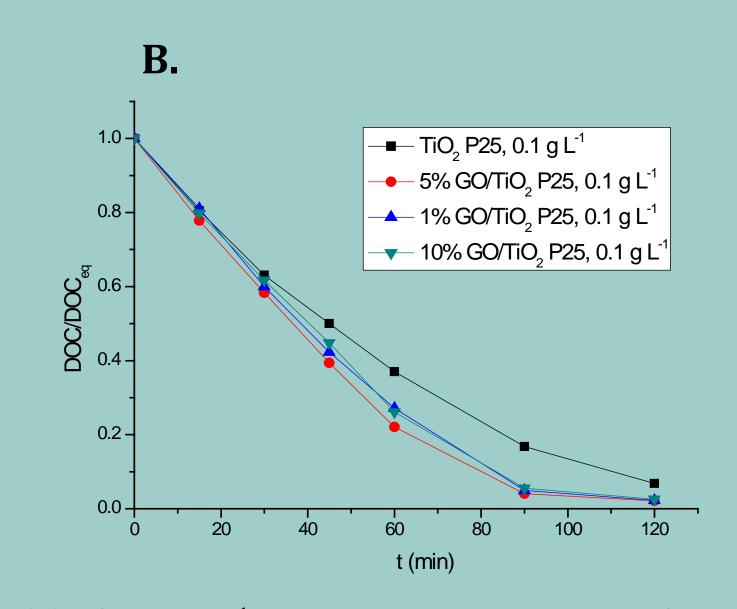




**Figure 2.** SEM of 1 and 5%  $GO/TiO_2$  P25 composites. A-B: 5  $\mu$ m, C-D: 10  $\mu$ m magnification.

**Figure 3.** FTIR specrta of graphene oxide and  $TiO_2$  P25 composites. Blue line: 1%  $GO/TiO_2$  P25, black line: 5%  $GO/TiO_2$  P25.





**Figure 4.** Photocatalytic oxidation (A) and mineralization (B) of 20 mg  $L^{-1}$  clopyralid in the presence of UV-A irradiation and 0.1 g  $L^{-1}$  GO/TiO<sub>2</sub> P25 composites.

## **Discussion and Conclusions**

- GO was successfully prepared from graphite according to the modified Hummers method. It was then, combined with commercial  $TiO_2$  P25 resulting to three different composites with 1, 5 or 10% GO content (nominal).
- SEM images revealed the successful preparation and complexation of GO with TiO<sub>2</sub>, demonstrating small sphere-like pTiO<sub>2</sub> particles smoothly deposited onto the GO sheets (Fig. 2).
- FTIR spectra reveal interesting findings about the prepared  $\rm GO/TiO_2$  P25 composites. First, a shift from 1600 cm<sup>-1</sup> to ~1640 cm<sup>-1</sup>, characteristic of the conjugation of the carbonyl groups of GO with functional groups on the surface of  $\rm TiO_2$ . Second, the slight difference in the transmittance between 1% and 5% sample at 3300 cm<sup>-1</sup> can be attributed to the different concentration of GO in the prepared mixed catalysts, which corresponds to different ratios of functional groups (-OH) in each sample (Fig. 3).
- In all cases, the initial degradation rates during photocatalytic degradation of the herbicide clopyralid, in the presence of UV-A irradiation, were enhanced in comparison to the bare TiO<sub>2</sub> P25 (Fig. 4A). Similarly, mineralization of the herbicide was faster under UV-A irradiation, when either of the three GO/TiO<sub>2</sub> P25 composites was employed (Fig. 4B).

## Acknowledgements

This research is implemented through the State Scholarships Foundation (IKY) program and cofinanced by the European Union (European Social Fund-ESF) and Greek national funds through the action "Reinforcement of Postdoctoral Researchers", in the framework of the Operational Program "Human Resources Development Program, Education and Lifelong Learning" of the National Strategic Reference Framework (NSRF) 2014-2020.

